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HARRY L. HOPKINS, Secretary
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of the

National Bureau of Standards

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Volume 22

January to June 1939

Research Papers 1163 to 1217

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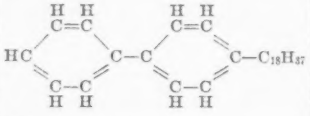
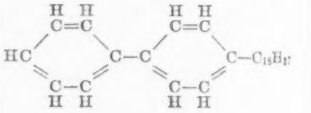
GOVERNMENT PRINTING OFFICE
WASHINGTON : 1939

Corrections to be noted in volume 22 of the Journal of Research of the National Bureau of Standards

Page	Line	Now reads in part	Should read
5	7	$(\delta-\delta'),(P-P')$	$(\delta-\delta')/(P-P')$
5	8	$P'/$	P',
5	Footnote 20		$\frac{w_n}{w_n}=\frac{1}{1-\frac{P}{P_n}}$
31	6 from bottom	$-(^2P)\ 6p\ ^2D^{\circ}_{3/2}$	$-(^2P)\ 6p\ ^2D^{\circ}_{1/2}$
39	3	$(^3P)\ 6s\ ^2P_{2/2}-$	$(^3P)\ 6s\ ^2P_{0/2}-$
75	Table 3, column 4.	$22.2\times10^{-}$	22.2×10^{-5}
91	1	*E*-layer	*F*-layer
127	12 of abstract	the difference	the deflection
202	Table 4, last column.	34	[a]34
231	24	elevation and $\pm\Delta c$	elevation and $\pm\Delta I_e$
237	17	table 5	table 4
275	Eq 4	$-T_{NaOH}.N_{NaOH}$	$-T_{NaOH}.N_{NaOH})$
283	Last line of par. 2.	fluidity, η, of the solution	fluidity, $1/\eta$, of the solution
316	15	$-CH_2,$	$>CH_2,$
316	15	$-CH,$	$\geqslant CH$
324	Formula 1		The double bonds should be single bonds.
324	Formula 3		
364	Table 5, last column.	3252	4252
376	17 from bottom.	695.87° C	659.87° C
386	Table 4, column 6.	−1912	−2912
292	14 of text	$I=(1.378$	$k=(1.380$
423	10 under VIII	3636.12	3536.12
473	Table 1, series 5.	solution=0.0997 g	solution=0.09997 g
478	4	attained	obtained
510	8th from bottom.	increase	decrease
510	10th from bottom.	decreased	increased
674	6 under II	1,800 μl	1,800 $\mu\mu l$

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